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Kinetic Studies of Zinc (II) Ion Incorporation into Polyvalent Porphyrin, Tetrakis(3,5-di-tert-butyl-4-hydroxyphenyl)porphyrin in Dimethylformamide*

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The incorporation of zinc (II) ion into porphyrins in dimethylformamide was studied kinetically. The porphyrins studied were tetrakis(3,5-di-tert-butyl-4-hydroxyphenyl)porphyrin and tetraphenylporphyrin. The rate of the reaction was measured spectrophotometrically. The rates under pseudo-first-order conditions were almost the same for the both porphyrins. The results showed that the bulky substituents in the meso-phenyl groups have little effect on the zinc (II) ion incorporation into the porphyrin ring. It has also been shown that the counteranions of the zinc (II) used have a substantial effect on the kinetics of the porphyrin metallation, and the compensation effects have been observed in the seven zinc (II) salts examined. On the basis of zinc concentration dependence and temperature dependence of the rate constants, metallation mechanism was discussed.

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