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# One Step Synthesis of $5,2^{\prime}$－Bithiazole Derivatives：Reaction of $\mathbf{N}, \mathbf{N}$－Dimethyl－2，4－dithiobiuret with $\alpha$－Haloketones 

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Thermal reaction of $\mathrm{N}, \mathrm{N}$－dimethy1－2，4－dithiobiuret（la）with an equivalent of $\alpha$－ haloketone， $\mathrm{X}-\mathrm{CH}_{2}-\mathrm{C}-\mathrm{R}$（ $2 \mathrm{a}: \mathrm{X}=\mathrm{Cl}, \mathrm{R}=\mathrm{CH}_{3}$ ），in aqueous solution gave 4－methy1－2－（ $\mathrm{N}, \mathrm{N}-$ dimethylthiocarbamoyl）aminothiazole（3a）and structurally unknown substance as pale yellow crystals（ $4 \mathrm{a}-\mathrm{HCl}$ ）（from $2-\mathrm{PrOH}$ ）， $\mathrm{mp} 196-198^{\circ}$ in equal amounts（yield， 90 $\%$ ）．Neutralization of（ $4 \mathrm{a}-\mathrm{HCl}$ ）with $\mathrm{NaHCO}_{3}$ solution gave a free base（4a），mp 85－ $86^{\circ}$（from hexane）．The free base（4a）was also obtained in $90 \%$ yield by the react－ ion of（3a）and（2a）．In the NMR spectrum（ $\mathrm{CDCl}_{3}$ ）of（4a）a signal at $\delta 3.13(3 \mathrm{H}, \mathrm{s})$ indicated the presence of $\frac{\mathrm{CH}_{3}}{\mathrm{CH}_{3}}>\mathrm{N}$－，a doublet at $\delta 2.42(3 \mathrm{H}, \mathrm{J}=1.0 \mathrm{~Hz})$ was due to the signal for $4-\mathrm{CH}_{3}$ which undergoes a long－range coupliug with $5-\underline{H}$ in the thiazole ring appearing at $\delta 6.72(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=1.0 \mathrm{~Hz})$ ，and a singlet at $\delta$
$2.50(3 \mathrm{H}, \mathrm{s})$ corresponded to the $\mathrm{CH}_{3}$ in the thiazole ring． $\mathrm{CH}_{3}$
These data suggest the following bithiazole structure for $\mathrm{CH}_{3}$ （4a）．
In order to establish unequivocally the bithiazole structure 4 －methyl－2－dimethyl－ amino－5－（4＇－methylthiazol－ $2^{\prime}-\mathrm{yl}$ ）thiazole（4a）was synthesized independently by the con－ densation of（2a）with 4－methyl－2－dimethylamino－5－thiocarbamoylthiazole（7），as shown in Scheme 1．IR and NMR spectral data of this compound agreed with those of the above－mentioned pale yellow crystals．Therefore，the structure of（4a）was finally established as $5,2^{\prime}$－bithiazole structure，4－methyl－2－dimethylamino－5－（4＇－methylthiazol－ $2^{\prime}$－yl）thiazole．The similar reaction of（1a）and（2b）gave only orange yellow precip－ itates（4b）in $90 \%$ yield and（3b）was not obtained．

However，reaction of（1a）and（2b）in dioxane at room temperature gave（3b），mp $168^{\circ}$（from 2－PrOH）in $80 \%$ yield．The elemental analysis and mass spectral data gave a molecular formula of $\mathrm{C}_{12} \mathrm{H}_{13} \mathrm{~N}_{3} \mathrm{~S}_{2}$ for 2－（ $\mathrm{N}, \mathrm{N}$－dimethylthiocarbamoyl）amino－4－phenyl－ thiazole（3b）．

In a previous paper，we reported that the reaction of N －methyl－2，4－dithiobiuret （1b）and（2a）gave 2－（4－methylthiazol－2－yl）imino－3，4－dimethyl－4－thiazoline（8）via route（A），in which a nucleophilic attack of the amide－nitrogen to the carbonyl－car－ bon in an intermediate（ 5 c ）is included．In contrast，（5a）and（ 5 b ）may be produced
by the condensation via route (B), in which an active methylene in the intermediate ( $5 \mathrm{a}, \mathrm{b}$ ) attacks on a carbon at 2 -position of the thiazole ring.


