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A NOVEL BIOSYNTHETIC STUDY OF GRISEOFULVIN BY ²H NUCLEAR MAGNETIC RESONANCE: DETERMINATION OF DEUTERIUM INCORPORATION FROM [2-²H₈]-ACETATE BY *PENICILLIUM URTICAE* *

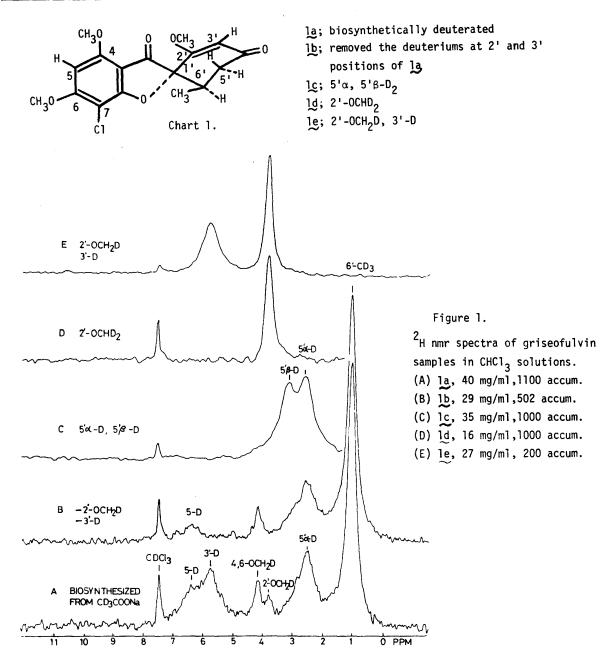
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In the elucidation of the biosynthetic pathways, the use of ¹³C nmr combined with ¹³C-label-precursors has been common practice to locate the enriched site and determine the skeleton-formation. This method, however, does not provide an unambiguous information on biosynthetic pathways involving hydrogen. For this purpose, the use of ²H nmr in cass of ²H-labeled precursors seems to have potential utility for the location of deuterium incorporation, together with mass-spectrometric analysis. Although very few works have been done on this subject partly because of fear of lower sensitivity and wider line-width of deuterium signal, recent developments of the pulsed Fourier transform nmr method have enabled us to study various types of ²H nmr to chemical and biological problems. We now wish to demonstrate that direct evidence of deuterium incorporation and its stereochemical course on biosynthesis of griseofulvin are obtained from ²H nmr when $(2-²H_3)$ -acetate is used as a tracer for the biosynthesis, which is in good agreement with the previous studies using (2-³H, ¹⁴C)-acetate.

²H nmr spectra were recorded by a JEOL PFT-100/EC-100 pulsed Fourier transform spectrometer operating at 15.28 MHz with proton-noise decoupling. All samples of chloroform solution were contained in 10 mm o. d. sample tubes. Field-frequency control was performed on the internal signal of C_6F_6 , which was added by amounts of a few drops in the chloroform solution. The biosynthetically deuterated griseofulvin (1a) was prepared from sodium $[2^{-2}H_3]$ -acetate by *Penicillium urticae* as previously reported. In order to perform unambiguous assignment of ²H signals, a series of selectively deuterated griseofulvin samples were prepared (Chart 1). In Figure 1A is shown a ²H nmr spectrum of biosynthetically deuterated griseofulvin (1a) in CHC1₃ solution (4 w/%). The lowermost sharp signal arises from CDC1₃ occuring in CHC1₃ of natural abundance (0.02%). The assignment of ²H nmr signals is straightforward to that of ¹H nmr, since chemical-shift displacement due to isotope effect are usually negligible. In this communication, however, the peak-assignments were made with the aid of ²H signals of selectively deuterated griseofulvin samples described above. First, the peaks of 2'-OCH₂D and 5-D are assigned by comparing ²H nmr spectrum of la with that of lb (Figure 1B), in which deuteriums are removed

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at 2'-methoxyl and 3'-position. The assignment of 2'-OCH₂D signal is also confirmed by employing [2'-OCH₂D, 3'-D]-griseofulvin (1e, Figure 1E) and [2'-OCHD₂]-griseofulvin (1d, Figure 1D) as the reference samples. In comparison with ²H nmr of 1c (Figure 1C), deuterium at 5'-position is confirmed to have been incorporated exclusively at α configuration. This result is in agreement with the previous studies on [2-³H, ¹⁴C]-acetate tracer. Further, ²H T₁ values show that deuteriums incorporated at methyl or methoxyl groups where internal rotation will be allowed in addiion to overall molecular tumbling are found to give larges T₁ values (86, 106 and 104 msec for 6'-CD₃, 2'-OCH₂D and 4, 6-OCH₂D, respectively) compared with 5' α -D and

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(45 and 46 msec, respectively).

In contrast to the case of ¹³C nmr, nuclear Overhauser enhancement by protondecoupling is negligible for ²H nuclei where the quadrupole relaxation mechanism is dominant. Accordingly, integrated peak-intensities are proportional to the extent of deuterium-incorporation by biosynthesis. The relative ²H peak-intensities of la are : 44% (6'-CD₃), 23% (5' α -D) 3.3% (2'-OCH₂D), 6.3% (4, 6-OCH₂D) and 24%(3'-D and 5-D). The comparison of the peak-intensities between 6'-CD₃ and 5' α -D strongly suggests that 6' position might be CHD_2 instead of CD_3 . This would be easily proved if doubling of the ²H signal due to geminal ²H-¹H spin coupling were observed. Unfortunately, no such a fine structure was observed in the proton-coupled ²H spectrum recorded under the condition of turning-off proton-decoupler. It is expected that this situation arises when peak-splittings due to ²H-¹H spin-couplings, (the splitting of which being 1/6 of corresponding ${}^{1}H^{-1}H$ couplings) are buried within relatively broader line-width. Employing $1/T_1$ as a theoretical limit of a line-width free from various broadening factors such as magnetic inhomogeneity and unresolved ²H-¹H spin-couplings, it is predicted that no fine structure could be observed unless otherwise $\pi T_1 J_{DH} \gg 1$. Here J_{DH} stands for ²H spin coupling constant. In our present case, $\pi T_1 J_{DH} \sim 0.5$ is obtained from the values of $T_1 \sim 100$ mesc and $J_{DH} \sim 1.7$ Hz. This value predicts that proton-decoupling experiment will alter spectral pattern to some extent. In fact, the peak heights of 2'-OCH₂D and 4. 6-OCH₂D are found to be increased by amounts of 27% and 21%, respectively, when compared with those Therefore, the enhanced peak-height of 6'-methyl by of proton-coupled spectrum. amount of 17% suggests that the deuterium-incorporation at 6'-position is apparently like CHD₂. Further, it is of interest to note that 9.6% of deuterium is incorporated at unexpected methoxyl groups $(4, 6- \text{ and } 2'-\text{OCH}_2\text{D})$. Such an analysis could not readily be performed by other physical techniques.

In conclusion, it is proved that ²H nmr is very powerful nondestructive method to study biosynthetic pathways involving hydrogen.